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The Temperature and  
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Gizmo™ illustrates  
how the molecules of  
gas move at different  
temperatures. In this  
Gizmo, temperature  
is measured on the

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Kelvin scale, which measures temperature from absolute zero, the coldest possible temperature ( $-273.15^{\circ}\text{C}$ ).

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Question: How is the temperature of a gas related to the motion of gas molecules? 1.

Observe: Move the Temperature slider back and forth. Focus on the particle motion at left. What do you notice? The colder it gets the slower they go the hotter it gets the faster they will go.

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Name: Anaya Tei

Date: October

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Exploration:

Temperature and  
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Vocabulary: absolute  
zero, Kelvin scale,

kinetic energy,  
Maxwell-Boltzmann  
distribution, molar

mass, molecule,  
temperature,

universal gas  
constant Prior

Knowledge

Questions (Do these



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BEFORE using the Gizmo.) 1. Why is hot air hot? Hot air is hot because the sun is radiating hot oxygen 2.

Science .pdf - Name

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The Temperature and Particle Motion

Gizmo™ illustrates

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how the molecules of gas move at different temperatures. In this Gizmo, temperature is measured on the Kelvin scale, which measures temperature from absolute zero, the coldest possible temperature ( $-273.15^{\circ}\text{C}$ ).

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2019 Name: \_\_\_\_\_

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Exploration:

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Vocabulary: absolute  
zero, Kelvin scale,  
kinetic energy,  
Maxwell-Boltzmann  
distribution, molar  
mass, molecule,

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Knowledge

Questions (Do these

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rticle\_Motion\_Gizmo.

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Knowledge

Questions (Do these BEFORE using the Gizmo.)

1. Why is hot air hot? Hot air rises

because when you heat air (or any other gas for that matter), it expands. When the air expands, it becomes less dense than the air around it.

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Vocabulary: absolute  
zero, Kelvin scale,  
kinetic energy,  
Maxwell-Boltzmann  
distribution, molar  
mass, molecule,  
temperature,  
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Knowledge Q.

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Gizmo.) Why is hot air

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In the Temperature

and Particle Motion

Gizmo, students

explore how the

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molecular weight of a gas relates to the distribution of particle velocities.

The Gizmo includes a simulation that shows how particles in a gas collide and how momentum and kinetic energy are transferred between particles.

Gizmo of the Week:

*Page 24/40*



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Particle Motion ...

Temperature and

Particle Motion

Observe the

movement of

particles of an ideal

gas at a variety of

temperatures. A

histogram showing

the Maxwell-

Boltzmann velocity

distribution is shown,

and the most

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probable velocity, mean velocity, and root mean square velocity can be calculated. Molecules of different gases can be compared.

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