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$p = f(t) + O(t^2)$  (1.2) Therefore, the contribution to the average electron momentum from the electrons that do not collide is  $p(t) = (1 - \tau/p)(t) + 1 + O(\tau)$  (1.3) Note that the contribution to the average momentum from electrons which have collided will be of order  $\tau$  (1.2) and therefore negligible.

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It has one unpaired electron in the 3s orbital. The 3s valence atomic orbital of sodium overlaps with another such orbital of the same energy to form molecular orbitals. The atomic orbitals continue to combine in some fashion forming a band.

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Fe (s) + CuSO 4 (aq) - FeSO 4 (aq) + Cu (s) Fe (s) + FeSO 4 (aq) - no reaction. Fe (s) + Al 2 (SO 4) 3 (aq) - no reaction. Fe (s) + ZnSO 4 (aq) - no reaction. No reaction occurs when any of the metals from zinc, iron, aluminium, and copper are placed in aqueous solution of Al 2 (SO 4) 3.

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