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charge experienced by the  
electrons in the  $n=3$   
electron shell: K, Mg, P,

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Rh, and Ti. Rank from  
smallest to largest. To rank  
items as equivalent, overlap  
them. Mg, P, K, Ti, Rh Part  
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Part A. Arrange the following atoms in order of increasing effective nuclear charge experienced by the electrons in the  $n=3$  electron shell: K, Mg, P, Rh, and Ti. Rank from smallest to largest. To rank items as equivalent, overlap them. Mg, P, K, Ti, Rh. Part A.

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