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Halogens have the highest negative value of electron gain enthalpy. Here Cl and F both are halogens but electron gain enthalpy of Cl is more negative than F because of the very small size of F which causes interelectronic repulsion among its electrons.

Which of the following will have the most negative ...

(b) C is less reactive than A.As both elements belong to same group and going down to group size of element increases. A is a smaller than C therefore the cloud of electron density surrounding the nucleus is more compact. The stability of completing the octet for A is greater than for C as A is more keen to obtain an electron than C.

The position of three elements A, B and C in the Periodic ...

Test 2 also consists of a Training section and an Exam practice section for each part of the exam. The Training sections are shorter than those in Test 1. They review the information provided in Test 1 and also include further practice for that part of the test. The Exam practice sections provide additional tips and advice.

Advanced Trainer - 6 Practice Tests [1q7j44e3nxqv]

Take A Sneak Peak At The Movies Coming Out This Week (8/12) Taking your child to a movie theatre for the first time; New Movie Releases This Weekend: April 16th – April 18th

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