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Osmium has slightly lower density (22.52 g cm⁻³) as compared to iridium (22.61 g cm⁻²).Thus, iridium has the highest density among transition metals. (viii) Atomic volume Atomic volume decreases along the period due to decrease in atomic size. (ix) Reactivity d-block elements are less reactive due to high ionisation energies. Some are almost inert and known as noble metals, e.g., Au; Pt, Os ...

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