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Chapter 11 - Thermochemistry - Heat and Chemical Change Chapter 11: 1 - 35, 57, 60, 61, 71 Section 11.1 - The Flow of Energy - Heat Practice Problems 1. When 435 J of heat is added to 3.4 g of olive oil at 21? C, the temperature increases to 85? C. What is the specific heat of olive oil? Knowns: q = 435 J; m olive oil = 3.4 g; ?T = (85 ?C - 21 ?C); Use C = q/ m ??T = 435 ? ...

Chapter 11 Thermochemistry Heat and Chemical Change

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Unit A: Thermochemical Changes - W. P. Wagner Science

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Chemistry chapter 11 Chemical reactions answer key. coefficient. a whole number that appears before a formula in an equation. spectator ion. a particle not directly involved in a chemical reaction. combustion reaction. a reaction in which oxygen reacts with another substance, often producing light or heat. reactant . a starting substance in a chemical reaction. skeleton equation. a chemical ...

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11-Thermochemistry 12-Colligative Properties 13-Thermodynamics 14-Gas Phase, Solubility, Complex Ion Equilibria 15-Kinetics 16-Acid-Base Equilibria 17-Electrochemistry < Previous: Chapter Eleven listing is 2 pages: Next > Page 1: Page 2 : Chapter Eleven - Thermochemistry: Videos: Gummy Bear Oxidation This video shows the oxidation of a gummy bear with molten KClO_3 . optional version ...

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